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4.8.The rate of the chemical reaction doubles for and increase of 10 K in absolute temperature from 298 K. Calculate E a. Ans. 4.9.The activation energy for the reaction, $2 \text{HI(g)} \rightarrow \text{H}_2 + \text{I}_2 \text{(g)}$ is 209.5 k J mol⁻¹ at 581

K. Calculate the fraction of molecules of reactants having energy equal to or greater than activation energy?

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