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Physics and Statistical Mechanics Alexandre V. Morozov 1 Solutions for Problem 1 a) $Q = 0$ for adiabatic processes, and thus the rst law of thermodynamics becomes: $U + A = 0$; (1) where A is the work done by gas, and U is its internal energy. Using $A = P \Delta V$ and $U = C_V \Delta T$ we obtain: $C_V(T_2 - T_1) + P(V_2 - V_1) = 0$: (2)

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