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$\text{Zn} + 2\text{H}^+ \rightarrow \text{Zn}^{2+} + \text{H}_2$
 $\text{NH}_3 + \text{H}_2\text{O} \rightarrow \text{NH}_4^+ + \text{OH}^-$
 $\text{HClO} + \text{LiOH} \rightarrow \text{LiClO} + \text{H}_2\text{O}$
 $\text{HCl} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{Cl}^-$
 pH Scale
 ranges from 0-14 Acids

found between a
 number close to 0 & 7
 Bases found between 7
 & 14 Neutral 7
 measures the acidity or
 basicity of a solution
 by focusing on the
 concentration of
 hydrogen ions (H^+) in
 a solution equals the
 negative log of the
 concentration of H^+
 Based on the prefixes
 "hetero" and "homo,"
 what do you think are
 characteristics of these
 two types of mixtures?

**Solutions - Science
 with Mr. Jones**

A solution is a special
 type of mixture.
 Solutions Solutions A
 solution is a
 homogeneous mixture
 that combines a solute
 and a solvent. A
 solution involves one
 substance dissolving in
 another substance,
 which is usually a
 liquid. 11. A solution is

a special type of mixture.

Mixtures & Solutions PPT - SlideShare

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EQ: What are Mixtures and Solutions? This is a bundled lesson that includes the student interactive notes and the Power Point

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as a supplement to their own class notes or as a review.

Mrs. J's Physical Science Page - Lecture Notes

Title: Pearson Prentice Hall Physical Science: Concepts in Action
Author: Diane Last
modified by: Diane
Created Date: 8/12/2008 8:01:35 PM
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PPT - Pearson Prentice Hall Physical Science: Concepts in ...

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on the amount of solute added to the solvent. A dilute solution contains a small amount of solute in a large amount of solvent. A concentrated solution contains a large amount of solute dissolved in a small amount of solvent.

Types of Solutions - Different Types, Homogeneous ...

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1- Skills of scientist ... –
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