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1- Iron reacts with oxygen to produce iron (III) oxide. $4 \text{ Fe (s)} + 3 \text{ O}_2 \text{ (g)} \rightarrow 2 \text{ Fe}_2 \text{ O}_3 \text{ (s)}$ If 7.18 moles of Fe react with excess O_2 , how many moles of $\text{Fe}_2 \text{ O}_3$ - The central atom in the triiodide ion, I_3^- , is surrounded by one single bond, one double bond, and one lone pair of electrons. two single bonds and no

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1) During a Chem 210 experiment, students will try to separate a mixture consisting of 0.50 g of salicylic acid (a solid) and 0.50 g

of triphenyl methanol (a neutral solid) using a technique called extraction.