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Solubility Rules. A solute is considered soluble if an appreciable amount of it can be dissolved in a given amount of the solvent. For example, both table salt (NaCl) and table sugar (C₁₁ H₂₂ O₁₁) are soluble substances in water. A solute is considered insoluble if very little of it dissolves in a given amount of the solvent. For example, sand (SiO₂) is considered insoluble in water.

5A: Solubility and Solution Reactions (Worksheet ...

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SOLUBILITY RULES Not all solutes dissolve to form solutions. Based on experimental observations chemists have drawn up a list of solubility rules for common salts when dissolved in water 1. All nitrates (NO₃⁻) and acetates are soluble. 2. All salts of the alkali metals and ammonia are soluble. 3.

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