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ESS55 Prof. Jin-Yi Yu The First Law of

Thermodynamics □ This law states that (1) heat is a form of energy that (2) its conversion into other forms of energy is such that total energy is conserved.

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### **Thermodynamics - Wikipedia**

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To be able to use the First Law of Thermodynamics to estimate the potential for thermo- mechanical energy conversion in aerospace power and

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established principle of thermodynamics (which eventually became the Second Law) was formulated by Sadi Carnot in 1824. By 1860, as found in the works of those such as Rudolf Clausius and William Thomson, there were two established "principles" of thermodynamics, the first principle and the second principle.

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