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of H_2 molecules = $2 \times 1 = 2$ g. (b) 1 mole of N atoms = 14 g

Textbook - Worked Solutions Chapter 4 - The Periodic Table ...

Solutions Manual Chemistry: Matter and Change • Chapter 11 211 CHAPTER 11 SOLUTIONS MANUAL Section 11.1 Assessment page 372 5 Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related The coefficients in the balanced equation indicate the molar relationship between each Chemistry Chang 11th Answers - test.rrammuseum.org.uk ...

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