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Activity Series Lab Microscale Answers

The purpose of the lab was to find which metal is the most reactive and which metal is the least reactive. It was known before the experiment that the metals used in the experiment are placed in the activity series from most active to least active as follows: magnesium, aluminum, zinc, and copper. The hypotheses formed were that zinc nitrate would react with aluminum and magnesium; aluminum nitrate would react with magnesium; copper nitrate would react with zinc, magnesium, and aluminum; and ...

Activity Series Lab Answers | SchoolWorkHelper

ACTIVITY SERIES LAB (MICROSCALE) Purpose: to study the chemical activity of common metals Safety Precautions Wear goggles at all times. Review precautions for handling acids. Since silver nitrate may stain skin and clothing, avoid contact. Procedure (Part I): 1. Sketch out a matrix to record your data. Along the top of the matrix, you should list

ACTIVITY SERIES LAB - Auburn School District

2Ag+ (aq) + Zn (s) 2Ag (s) + Zn+2 (aq) The list in decreasing ease of oxidation for this data is: Mg, Zn, Pb, Cu, Mg. The activity series for these metals are: Mg, Zn, Pb, Cu, Ag. Direct correlation.

An Activity Series Lab by av s - Prezi

Grade 11 Chemistry Activity Series of Metals Lab Problem: What is the order of reactivity of the metals copper, iron, magnesium, and zinc in single displacement reactions? Materials: Wellplate/Spotplate Small pieces of magnesium, iron, zinc and copper metal Dilute solutions of hydrochloric acid, copper (II) sulfate, zinc chloride, magnesium chloride, iron (III) sulfate Wash bottle with ...

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Question: EXPERIMENT 6: RELATIVE REACTIVITIES OF METALS AND THE ACTIVITY SERIES Name: Instructor: Post-Lab Instructor When You Have Completed The Report.) PART A & B: REACTION WITH WATER Date: Section/Group: Report (Use The In-lab Observations To Complete The Laboratory Report. Turn In To Your Als That Reacted With Water In Order Of Decreasing Reactivity (most ...

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TCU Department of Geological Sciences

Activity Series Lab Answer Key PDF Download - OsvaldAjeet ACTIVITY SERIES LAB (MICROSCALE) Purpose: to study the chemical activity of common metals Safety Precautions Wear goggles at all times. ... Answers to questions in complete sentences . REVISED 12/2003 Activity Series Lab – Observations

Activity Series Chemistry Lab Answers

When an atom gains electrons, it is reduced. Metals higher on the activity series are more likely to reacts relative to those lower on the activity series. The activity series can be used to predict products of reactions, and to predict if a reaction will even occur. In this experiment, different metals were tested for their reactivity. It was recorded if a reaction occurred or not, so that an activity series could be created. Data & Results

Chemistry Lab Report (The Activity Series) - Sarah Jackson

the more reactive nonmetals will displace ions of the less reactive halides from solution. In an activity series of nonmetals, the most reactive halogen is the one most easily reduced. The first part of this experiment derives an activity series for metals and uses a microscale technique. The second part derives an activity series for halogens.

AP Chemistry Lab 3 1 Activity Series of Metals and Nonmetals

Kaitlyn Kerkhoff Chem 161L Sec 10 March 19, 2012 Activity Series Lab Purpose: The purpose of this lab is to determine and observe the reactivity of metals with certain acids and solutions and from their derive the net ionic equations. From here we can determine which metal is the highest and lowest reactivity. Materials: • 7 test tubes • HCl solution • Pieces of Ca, Cu, Fe, Mg, Sn and Zn ...

Activity Series Lab - Kaitlyn Kerkhoff Chem 161L Sec 10 ...

The activity series allows us to predict whether a metal displacement reaction will occur. ... Answer. Yes. Magnesium is above copper on the reactivity series of metals. Therefore, it will replace the copper in the copper chloride, producing magnesium chloride and solid copper.

Metal Activity Series - Chemistry | Socratic

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Developing the Activity Series. Overview: The students will develop their own activity series of metals based on lab results. Qualitative observations will be used. Tie-ins include single-replacement reactions (SRR), oxidation-reduction, practical applications such as galvanization, and predicting reactions. Students work in

Developing the Activity Series - ASM International

In the Penny-Ante Equilibrium: A Classroom Activity—ChemTopic™ Lab Activity, pennies are used as reactants and products in a reversible reaction to answer questions about the fundamental nature of equilibrium and what happens to the amount of reactants and products when it is reached.

Student Lab Activities - Flinn Scientific

An Activity Series of Ions Lab Report Shakil M, Horby L, Shaquile P, Fazal M April 11th 2012 Mr. Vu Chemistry Introduction: Many reactions occur in our daily lives to benefit and improve our wellness and surroundings.

An Activity Series of Ions Lab | Magnesium | Zinc | Free ...

Play this game to review Chemical Reactions. Based on the activity series, will this reaction occur? Ni (s) + H 2 O (l) →