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2 • Quantum Mechanics: 500 Problems with Solutions as quanta of energy $h \nu$, where ν is the frequency of radiation. The individual quanta of light are called photons. Einstein’s photoelectric equation $1/2 m v^2 = h \nu - \phi$ (1.4) explained all aspectsof photoelectric effect. In Eq. (1.4), $h \nu$ is the energy of the incident photon, $h \nu_0$

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