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number of valence
electrons in each of the
following elements. a.

Ne d. Mg f. Cl h. Si 2.

Identify the energy
level of the valence
electrons in each of the
following elements. ...

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1. For each of the
following pairs of
elements, underline
the one that would
replace the other
element in a
compound. a. calcium,
tin e. iron, copper b.
bromine, fluorine f.
iodine, chlorine c.
aluminum, potassium
g. silver, lead d. zinc,
sodium 2.

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Teaching Transparency
Worksheet Answers
Chapter 6 math skills

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MATH SKILLS TRANSPARENCY WORKSHEET 2
Interpreting and Drawing Graphs Use with Chapter 2, Visualizing the Use

with Chapter 3, Section 2.4 Conservation of Mass Section 3.2 1.

MATH SKILLS TRANSPARENCY MASTER Visualizing the ... Pages ...

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 9 11 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

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Worksheets

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TEACHING
 TRANSPARENCY
 WORKSHEET Atomic
 Orbitals 1. What is the
 shape of an s orbital?
 Class 16 Use with
 Chapter 5, Section 5.2
 2. What is the
 relationship between

the size of an s orbital
 and the principal
 energy level in which it
 is found? 3. What is the
 shape of a p orbital?
 How many p orbitals
 are there in a ublevel?
 4.

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