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CHAPTER 2 REVIEW Measurements and Calculations SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Complete the following conversions: a. 100 mL 0.1 L b. 0.25 g 25 cg c. 400 cm³ 0.4 L d. 400 cm³ 0.0004 m 2. For each measuring device shown below, identify the

quantity measured and tell when it would remain constant ...

2 Measurements and Calculations

CHAPTER 2 REVIEW Measurements and Calculations SECTION 2-3 SHORT

ANSWER Answer the following questions in the space provided. 1. Report the number of significant figures in each of the following values: a. 0.002 37 g d. 64 mL b. 0.002 037 g e. 1.3 102 cm c. 350. J f. 1.30 102 cm 2. Write the value of the following operations using scientific notation: a. $103\ 1\ 0\ 1\ 2\ 0\ 6$ b. $8\ 2\ 1\ 1\ 0$

2 Measurements and Calculations

2(g) $\rightarrow$ NO 2(g) a. Balance the equation. $2\text{N}_2\text{O(g)} + 3\text{O}_2\text{(g)} \rightarrow 4\text{NO}_2\text{(g)}$ 4 mol NO 2:3 mol O 2 b. What is the mole ratio of NO 2 to O 2? 15.0 mol c. If 20.0 mol of

NO 2 form, how many moles of O 2 must have been consumed? True d. Twice as many moles of NO 2 form as moles of N 2O are consumed. True or False? False e. Twice as many grams of NO 2 ...

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PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 8. 9.7×10^{14} Hz What is the frequency of light that has a wavelength of 310 nm? 9. 9.4×10^{10} m What is the wavelength of electromagnetic radiation if its frequency is 3.2×10^2 Hz? Wavelength Wavelength Wave A Wave B 26 ARRANGEMENT OF ELECTRONS IN ATOMS ...

4 Arrangement of Electrons in Atoms

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CHAPTER 6 REVIEW Chemical Bonding
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

6 Chemical Bonding

CHAPTER 5 REVIEW The Periodic Law
SECTION 1 SHORT ANSWER Answer the following questions in the space

provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

Modern Chemistry CHAPTER 9
HOMEWORK 9-1 . Modern Chemistry
CHAPTER 9 HOMEWORK 9-2 (pp.
280282) .. write the correct symbols or
formulas for the word equa nons and . 5.
aluminum + . modern chemistry chapter
8 homework 8-2 (pp.. p.5: Section 2
Review: p.14: Section 3 Review: . Now is
the time to redefine your true self using
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Modern Chemistry Quiz . Name Section Quiz, continued Class Date 6. An experiment was designed to measure the effect of sulfur dioxide ... each sentence or best answers each question.
1. A chemical reaction was carried out three times. The mass of the product was 8.93 g for the first trial, 8.94 g for the second trial, and 8.92 g for the third ...

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Modern Chemistry 105 Chapter Test
Name Class Date Chapter Test A,

continued Use this figure to answer questions 7 and 8. ____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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