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SB.2. Question 29 - 2011 HSC Paper

SB.1. Question 27 - 2003 HSC Paper

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2012 HSC Chemistry 'Sample Answers'

Mass of lead in the precipitate = (207.2/278.2) X 0.595 = 0.443 g. Concentration of lead in the solution = 0.443 / 0.050 = 8.86 g/L. B is obtained by taking the formula of lead chloride as PbCl₂. C calculates the concentration of lead chloride in the solution (0.595 / 0.050). It is not obvious how D was obtained.

Conq Chem Website 2012 HSC exam answers

2012 HSC Chemistry Marking Guidelines. Section I, Part A. Multiple-choice Answer Key Question Answer. 1 B 2 D 3 A 4 C 5 B 6 C 7 D 8 D 9 B 10 B 11 B 12 D* 13 D 14 A 15 C 16 A 17 C 18 C 19 A 20 A * D was the best answer for Question 12. However, the correct answer is. 2-chloro-3-fluorobutane.

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2012 Chemistry GA 1: Written examination 1 GENERAL COMMENTS This examination was the final Unit 3 June examination for the VCE Chemistry Study Design. From 2013, a single examination covering both Units 3 and 4 will be held in November. Overall, students performed very well on the 2012 June Chemistry examination.

2012 VCE Chemistry 1 examination assessment report

2012 Trial HSC Examination HSC Chemistry General Instructions –Reading time 5 minutes –Working time 3 hours –Write using blue or black pen Draw diagrams using pencil Board-approved calculators may be used Write your student number at the top of each page where relevant A data sheet and Periodic Table are

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Answer: Explanation: 1: B: The UV protective ozone layer is located in the stratosphere. 2: D: A

burette is used for this purpose in a titration. The known volume of the other solution has been measured using a pipette and transferred to a conical flask. 3: B: Copper ions give a blue-green flame test. 4: C: Fe^{2+} reacts to form Fe^{3+} .

2015 HSC Chemistry Worked Solutions (Multiple Choice)

CHEMISTRY Wednesday, June 20, 2012 — 1:15 to 4:15 p.m., only This is a test of your knowledge of chemistry. Use that knowledge to answer all questions in this examination. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry.

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