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1. Test each solid with a conductivity tester. 2. Place the solid in water to determine if it forms a solution. 3. Test the solution with a conductivity tester to determine if it conducts an electric current. 4. Use a hammer to strike the solid and record your observations. Metallic solids conduct an electric current in the

Ionic Compounds and MetalsIonic Compounds and Metals

and Changes Chemistry - Chapter 9 Assessment Answers 91 Assessment 3 The positive charge of a group A metal is equal to its group number The charge of a group A nonmetal is the group number minus 8 4 from the number of electrons lost 5 -ite or -ate 6 Chapter 25 Assessment Chemistry Answers Chapter 25 Assessment Chemistry Answers Author:

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Prentice Hall Chemistry Chapter 7 Assessment Answer Key

General Chemistry I ANSWER KEY Problem test Chapter 7 Spring 1999 . 1. Show the electron configuration of all the electrons ,then show the electron configuration of the highest energy electron using the quantum number technique for each of the following atoms.

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7 Chemical Formulas and Chemical Compounds

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